

### mass transfer b k pdf

1.1 Molecular Mass Transfer For a binary mixture of A and B, the molar flux,  $N_{A,z}$ , of species A relative to the z axis is  $N_{A,z} = -cD \frac{dy_A}{dz} + y_A(N_{A,z} + N_{B,z})$  (1.1) In this equation, c is the total molar concentration,  $D_{AB}$  is the diffusivity of A in B,  $y_A$  is the mole fraction of A, and  $N_{B,z}$  is the molar flux of B.

### Chapter 1 Fundamentals of Mass Transfer

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mass transfer will be used throughout this module. 10.1 Mass transfer through diffusion . In Module 2 "Conduction", the Fourier equation was introduced, which relates the heat transfer to an existent temperature gradient:  $q = -k \frac{dT}{dx}$  (Fourier's law) (10.1) For mass transfer, where a component A diffuses in a mixture with a component

### MASS TRANSFER - NPTEL

mass transfer operation because the fluid adheres to the solid surface instead of dissolving in the solid. A desorption involves the transfer of mass from the solid surface (adsorbents) to the gas or liquid medium (adsorbates). A few operations of practical

### Mass transfer Operation - NPTEL

The principles of mass transfer, both diffusional and convective have been comprehensively discussed. The application of these principles to separation processes is explained. The more common separation processes

used in the chemical industries are individually described in separate chapters.

## **PRINCIPLES OF MASS TRANSFER AND SEPERATION PROCESSES**

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